DAMIETTA UNIVERSITY

CHEM-103:

BASIC ORGANIC CHEMISTRY

LECTURE 9

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(1) What is the IUPAC name for the following compound?



- (a) 1,3-pentamethylpropane
- (b) 1,1,3,3-tetramethylbutane
- (c) 2,4,4-trimethylpentane
- (d) 2,2,4-trimethylpentane

(2) When a double bond is formed between two atoms, one of the bonds is a sigma bond and the other is a pi bond. The pi bond is created by the overlap of...

- (a) sp² hybrid orbitals
- **(b)** sp³ hybrid orbitals
- (c) p orbitals
- (d) s orbitals

(3) What is the major product of the following reaction? ..



(4) The balanced equation for the combustion of pentane is

(a) $4 C_5 H_{12} + 2 O_2 \rightarrow 20 CH_4 + 4 H_2 O_2$ (b) $C_3 H_8 + O_2 \rightarrow 3 CH_4 + 2 H_2 O$ (c) $C_5 H_{12} + 8 O_2 \rightarrow 5 CO_2 + 6 H_2 O$ (d) $C_3 H_8 + 5 O_2 \rightarrow 3 CO_2 + 4 H_2 O$ (5) What are the major alkenes formed in the following reaction?



(6) Which product is obtained upon the reaction of (*Z*)-2-hexene with HBr?

(a) 2,3-Dibromohexane

(b) 2-Bromohexane

(c) 3-Bromohexane

(d) 2-Bromohexane and 3-bromohexane

(7) Which of the following compounds gives 3methyl-2-butanol upon reduction with LiAlH₄?



- (a) a
- (b) b
- (c) c
- (d) d

(8) Which of the following sets of reaction conditions will give the product indicated?



(a) $(CH_3O)_2Ag$, heat

(b) CH_3OH , HO^- , separate water

(c) 1) CH₃ONa, 2) Mel

(d) CH_3OH , H^+ , separate water

(9) In which molecule is the central atom *sp*³ hybridized?

A) CH_4 B) NH_3 C) H_2O D) All of these E) None of these

(10) What is the product of the following reaction?



(11) When benzaldehyde reacts with HCN and catalytic amounts of NaCN in ethanol, the product is



Homework: choose the correct answer

(12) For the following two pairs of molecules, identify which partner will have the higher boiling point.

